Research Article

Synthesis, Chemistry of [£]-allyl-^o-(^w, [£]-dimethoxyphenyl)- $H^{\gamma}, \tilde{f}, \tilde{f}$ -triazol- \tilde{f} -thiol

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Abstract

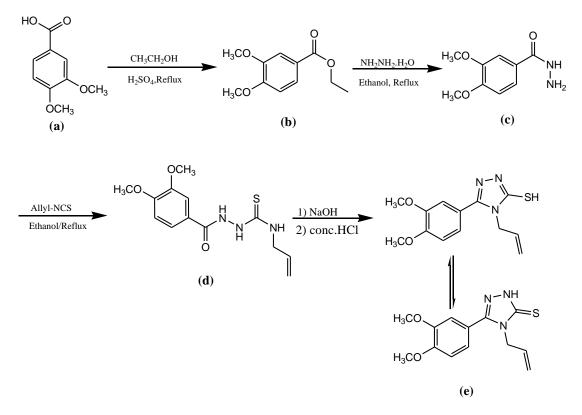
Triazole ring moiety has a wide spectrum of biological activities. The synthesis, chemistry of ξ -allyl- \circ -($(\pi, \xi$ -dimethoxyphenyl)- ξ H- $(\pi, \xi$ -triazol- $(\pi, \xi$ -triazol- (π, ξ) -dimethoxyphenyl)- ξ H- (π, ξ) -triazol- (π, ξ) -triazol- (π, ξ) -dimethoxyphenyl)- ξ H- (π, ξ) -triazol- (π, ξ) -triazol- (π, ξ) -dimethoxyphenyl)- ξ H- (π, ξ) -triazol- $(\pi,$ **Key word:** γ , ζ -triazole, allylisothiocyanate,

Introduction

Due to the different biological activities of $1, 7, \epsilon$ -triazole such as antimicrobial', antiinflammatory, antioxidant[°], anticonvulsant⁴, antitubercular[°] and anticancer[°] activity. The utility of these compounds not only in biological activity but also in synthesis of other heterocyclic compounds thiazolotriazoles^v, for examples: triazolothiadiazoles[^], triazolothiazines¹, and triazolothiadiazines'. There are several reported method for synthesis of triazole as Abdel Aziz H.A "reaction, Ueda" method and Einhorn-Brunner^{'''} synthesis.

Results and discussion

 ε -dimethoxyphenyl)- ε H-٤-Allyl-٥-(٣, ۱,^۲,^٤-triazol-^۳-thiol was prepared according to Scheme \. Esterification of r, ϵ -dimethoxybenzoic acid according to Fischer esterification' using ethanol in presence of conc. $H_{Y}SO_{\xi}$, treatment of ester **b** with hydrazine monohydrate 90% give hydrazide c. Reaction of hydrazide c with allyl isothiocyanate in presence of absolute ethanol afforded the corresponding carbazide **d**. Refluxing of carbazide **d** with ^YN NaOH followed by acidification with conc.HCl afforded the target compound e.



Scheme : Synthesis of ξ -allyl- \circ -($(\tau, \xi$ -dimethoxyphenyl)- ξ H- $(\tau, \xi$ -triazol- $(\tau, \xi$ -triazol- (τ, ξ) ٧٦ Synthesis, Chemistry of $\frac{1}{2}$ -allyl- $\frac{1}{2}$ -($\frac{1}{2}$, $\frac{1}{2}$ -dimethoxyphenyl)- $\frac{1}{2}H$ - $\frac{1}{2}$, $\frac{1}{2}$ -triazol- $\frac{1}{2}$ -

thiol

In the present study, the $1, 5, \epsilon$ -triazole-5-thiol derivatives was prepared by intramolecular cyclization of the $1, \epsilon$ -disubstituted thiosemicarbazides.

Experimental

chemistry

Chemical reactions were monitored by TLC, using Merck $\P^{\intercal} \land \circ$ pre-coated aluminum plate silica gel (Kiesel gel $\uparrow \cdot$) \circ cm× $\uparrow \cdot$ cm plates with a layer thickness of \cdot . \uparrow mm. The spots were detected by exposure to UV-lamp at $\lambda = \uparrow \circ t$ nm. Melting points were determined on Stuart electrothermal melting point apparatus and were uncorrected. NMR spectra ($\circ \cdot \cdot$ MHz for 'H, $\uparrow \uparrow \circ$ MHz for 'C) were observed in CHCl_r on Bruker AM $t \cdot \cdot$ spectrometer with tetramethylsilane as the internal standard. Splitting patterns are designated as follows: s, singlet; d, doublet; m, multiplet.

General procedure for synthesis synthesis of $\frac{1}{2}-\text{allyl-}^{\circ}-(\overline{v},\frac{1}{2}-\frac{1}{2})-\frac{1}{2}H^{-1},\overline{v},\frac{1}{2}-\frac{1}{2}H^{-1},\overline{v},\frac{1}{2}-\frac{1}{2}H^{-1}$ thiol.

Heating at reflux of benzoic r, ϵ dimethoxybenzoic acid with ethanol in the presence of concentrated sulphuric acid as a dehydrating agent afforded the corresponding esters. Hydrazinolysis of the ethyl ester derivatives with hydrazine monohydrate in refluxing ethanol afforded the corresponding carbohydrazides. The structure of the formed' hydrazides was confirmed by their reported melting points.

Heating at reflux of equimolar amounts of the hydrazides and allyl isothiocyanate in ethanol afforded the corresponding $, \xi$ disubstituted thiosemicarbazides which was used as a crude products for the next step.

Allyl-°-(",^t-dimethoxyphenyl)-^tH-¹,^t,^ttriazol-"-thiol.

White crystal, $\cdot . f^{\prime} gm$, $VA. q \cdot Z$; m.p: $f^{\prime} z$ $f^{\prime} \sigma OC(\text{Reported} f^{\prime} f^{\prime} - f^{\prime} f^{\prime} OC)^{\prime}$, 'H NMR $(\circ \cdot \cdot MHz, CDCl_{r}) \delta = f^{\prime} q^{\prime} (f^{\prime} H, s, - OC\underline{H}_{r}), f^{\prime} q \circ (f^{\prime} H, s, OC\underline{H}_{r}), \xi \cdot (q - \xi \cdot f^{\prime}) (f^{\prime} H, m, NC\underline{H}_{r}), \xi \cdot A^{\circ} (f^{\prime} H, d, J_{\text{trans}} = f^{\prime} \cdot f^{\prime} \cdot Hz, N-CH_{r}CH = C\underline{H}_{r}), \circ \xi \cdot (f^{\prime} H, d, J = f^{\prime} \cdot \xi \cdot Hz,$ N-CH_YCH=C \underline{H}_{Y}), \circ . $9 \le -7...$ (¹H, m, N-CH_YC<u>H</u>=CH_Y), 7..99 (¹H, d, J= \wedge .r· Hz, Ar- \underline{H}), \vee .19 (¹H, d, J= \wedge .r· Hz, Ar- \underline{H}), \vee .1r (¹H, s, Ar- \underline{H}), 11... (¹H, s, N \underline{H}); ¹C-NMR(17 \circ MHz, CDCl_r) $\delta \le 7...$ \circ \circ $7... \circ$, \circ 7... (¹T. 117...) 117... 1

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